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[Chapter 19 Acids Bases Salts](#)

Lakhmir Singh Solutions Class 10 Chemistry Chapter 2 Acids, Bases and Salts On the basis of the chemical property of compounds, it has been divided into three groups Acids, bases and salts where acids and bases combined to form salts.

[Acids Bases and Salts Class 10 Notes Science Chapter 2](#)

Mineral Acids - Acids that prepare from mineral are known as mineral acids/ inorganic acids/man-made acids or synthetic acid, such as hydrochloric acid (HCl), sulphuric acid (H₂SO₄), nitric acid (HNO₃), etc. Note: Organic acids are always weak but minerals acids can be strong as well as weak. We can take dilute weak acids (like organic acid) in our body.

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Inorganic compounds essential to human functioning include water, salts, acids, and bases. These compounds are inorganic; that is, they do not contain both hydrogen and carbon. Water is a lubricant and cushion, a heat sink, a component of liquid mixtures, a byproduct of dehydration synthesis reactions, and a reactant in hydrolysis reactions.

[14.4 Hydrolysis of Salts - Chemistry 2e | OpenStax](#)

Inorganic acids and inorganic oxygen compounds of non-metals; Halogen or sulphur compounds of non-metals; Inorganic bases and oxides, hydroxides and peroxides of metals; Salts and peroxysalts, of inorganic acids and metals; Miscellaneous; 2019-09-01: HTML PDF (95 KB) Chapter 29 - Organic chemicals